

Principles – Radioactivity (1)

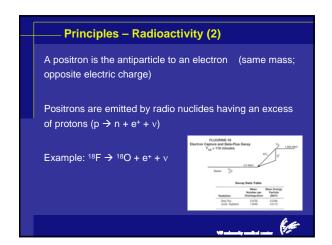
Radioactivity originates in a nuclear process caused by nuclear instability

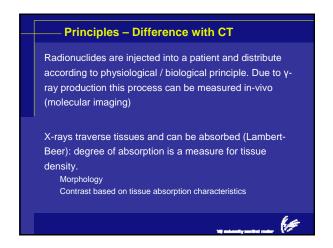
For a sample with N radioactive atoms the average decay rate is given by

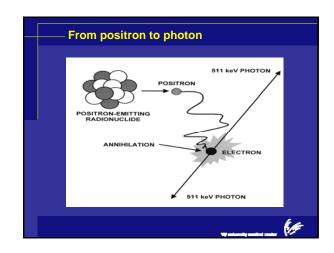
ΔN/Δt = -λN → N(t) = N(0) e^{λt}
with λ the decay constant for the radionuclide

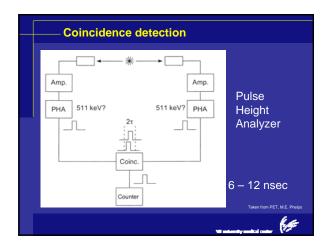
Units: A (Bq) = | ΔN/Δt | = λN

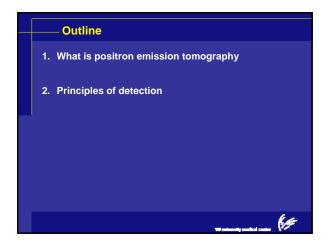
Imaging based on γ-rays (photons)

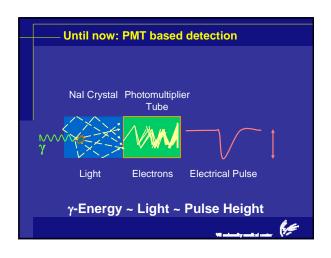


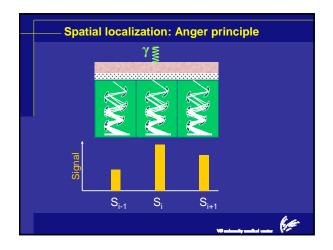


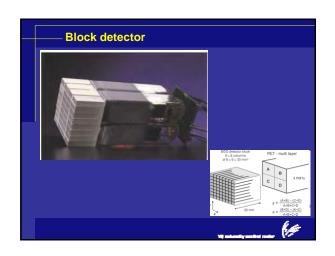


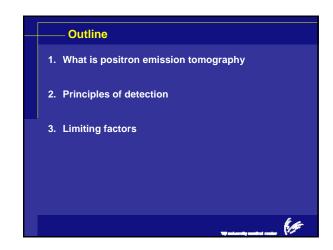


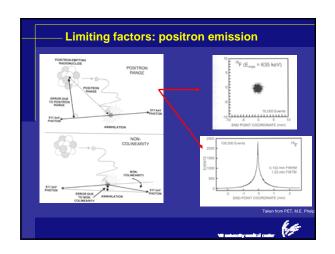


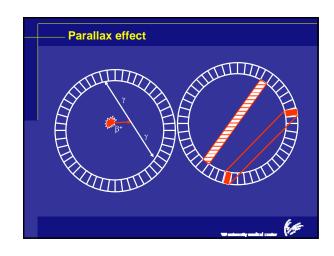


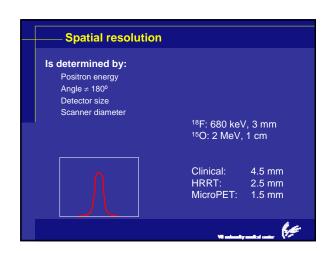


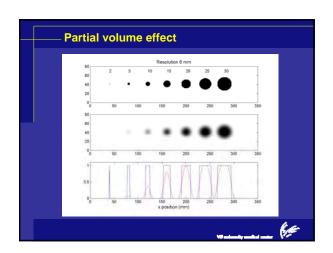


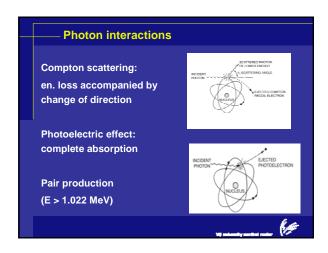


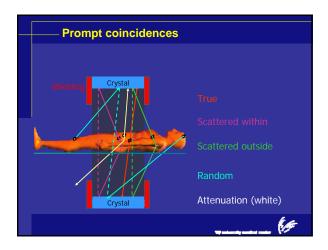


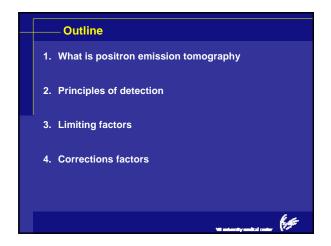


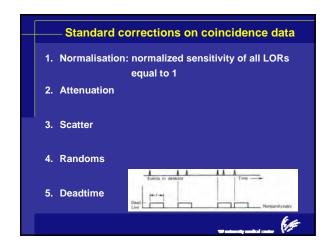


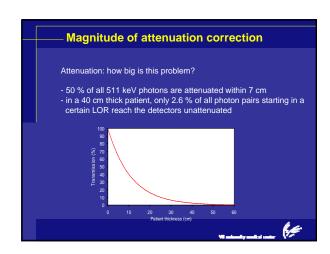


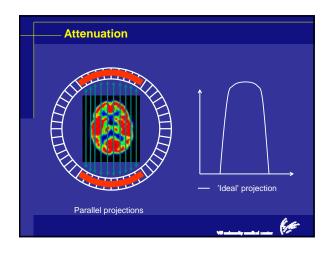


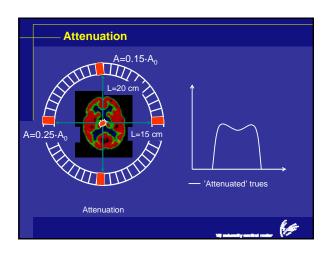


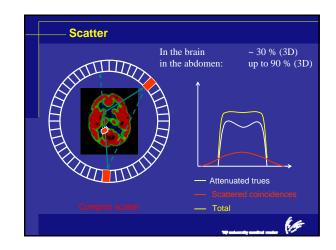


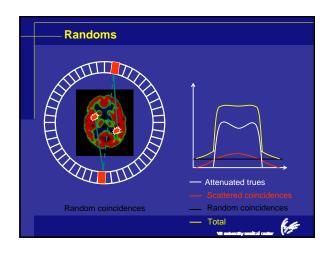


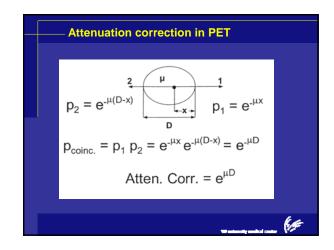


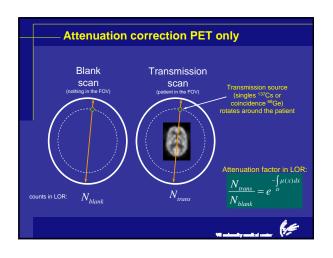


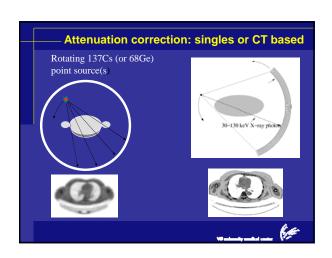


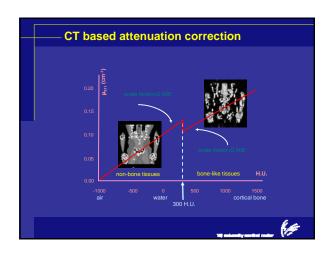


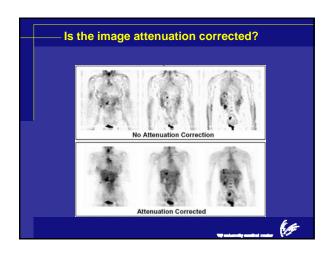


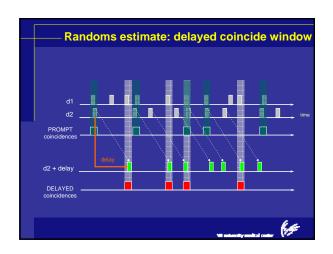


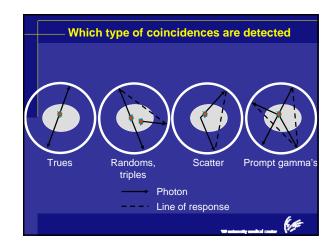












Standard corrections on coincidence data

1. Normalisation

2. Attenuation

3. Scatter

4. Randoms

5. Deadtime

Outline

1. What is positron emission tomography

2. Principles of detection

3. Limiting factors

4. Corrections factors

5. Quantification and quality control

